

Application of Nitric Acid Modified *Cocos nucifera*, *Pennisetum glaucum* and *Sorghum bicolor* Activated Carbon for Adsorption of H₂S Gas

Authors : Z. N. Ali, O. A. Babatunde, S. Garba, H. M. S. Haruna

Abstract : The potency of modified and unmodified activated carbons prepared from shells of *Cocos nucifera* (coconut shell), straws of *Pennisetum glaucum* (millet) and *Sorghum bicolor* (sorghum) for adsorption of hydrogen sulphide gas were investigated using an adsorption apparatus (stainless steel cylinder) at constant temperature (ambient temperature). The adsorption equilibria states were obtained when the pressure indicated on the pressure gauge remained constant. After modification with nitric acid, results of the scanning electron microscopy of the unmodified and modified activated carbons showed that HNO₃ greatly improved the formation of micropores and mesopores on the activated carbon surface. The adsorption of H₂S gas was found to be highest in modified *Cocos nucifera* activated carbon with maximum monolayer coverage of 28.17 mg/g, and the adsorption processes were both physical and chemical with the physical process being predominant. The adsorption data were well fitted into the Langmuir isotherm model with the adsorption capacities of the activated carbons in the order modified *Cocos nucifera* > modified *Pennisetum glaucum* > modified *Sorghum bicolor* > unmodified *Cocos nucifera* > unmodified *Pennisetum glaucum* > unmodified *Sorghum bicolor*.

Keywords : activated carbon adsorption, hydrogen sulphide, nitric acid, modification, stainless steel cylinder

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