

Removal of VOCs from Gas Streams with Double Perovskite-Type Catalyst

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Abstract : Volatile organic compounds (VOCs) are one of major air contaminants, and they can react with nitrogen oxides (NO_x) in atmosphere to form ozone (O₃) and peroxyacetyl nitrate (PAN) with solar irradiation, leading to environmental hazards. In addition, some VOCs are toxic at low concentration levels and cause adverse effects on human health. How to effectively reduce VOCs emission has become an important issue. Thermal catalysis is regarded as an effective way for VOCs removal because it provides oxidation route to successfully convert VOCs into carbon dioxide (CO₂) and water (H₂O(g)). Single perovskite-type catalysts are promising for VOC removal, and they are of good potential to replace noble metals due to good activity and high thermal stability. Single perovskites can be generally described as ABO₃ or A₂B₂O₄, where A-site is often a rare earth element or an alkaline. Typically, the B-site is transition metal cation (Fe, Cu, Ni, Co, or Mn). Catalytic properties of perovskites mainly rely on nature, oxidation states and arrangement of B-site cation. Interestingly, single perovskites could be further synthesized to form double perovskite-type catalysts which can simply be represented by A₂B'B''O₆. Likewise, A-site stands for an alkaline metal or rare earth element, and the B' and B'' are transition metals. Double perovskites possess unique surface properties. In structure, three-dimensional of B-site with ordered arrangement of B'O₆ and B''O₆ is presented alternately, and they corner-share octahedral along three directions of the crystal lattice, while cations of A-site position between the void of octahedral. It has attracted considerable attention due to specific arrangement of alternating B-site structure. Therefore, double perovskites may have more variations than single perovskites, and this greater variation may promote catalytic performance. It is expected that activity of double perovskites is higher than that of single perovskites toward VOC removal. In this study, double perovskite-type catalyst (La₂CoMnO₆) is prepared and evaluated for VOC removal. Also, single perovskites including LaCoO₃ and LaMnO₃ are tested for the comparison purpose. Toluene (C₇H₈) is one of the important VOCs which are commonly applied in chemical processes. In addition to its wide application, C₇H₈ has high toxicity at a low concentration. Therefore, C₇H₈ is selected as the target compound in this study. Experimental results indicate that double perovskite (La₂CoMnO₆) has better activity if compared with single perovskites. Especially, C₇H₈ can be completely oxidized to CO₂ at 300°C as La₂CoMnO₆ is applied. Characterization of catalysts indicates that double perovskite has unique surface properties and is of higher amounts of lattice oxygen, leading to higher activity. For durability test, La₂CoMnO₆ maintains high C₇H₈ removal efficiency of 100% at 300°C and 30,000 h⁻¹, and it also shows good resistance to CO₂ (5%) and H₂O(g) (5%) of gas streams tested. For various VOCs including isopropyl alcohol (C₃H₈O), ethanal (C₂H₄O), and ethylene (C₂H₄) tested, as high as 100% efficiency could be achieved with double perovskite-type catalyst operated at 300°C, indicating that double perovskites are promising catalysts for VOCs removal, and possible mechanisms will be elucidated in this paper.

Keywords : volatile organic compounds, Toluene (C₇H₈), double perovskite-type catalyst, catalysis

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