

## Nanostructure of Gamma-Alumina Prepared by a Modified Sol-Gel Technique

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**Abstract :** Nanoporous  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> samples were synthesized via a sol-gel technique, introducing changes in the Yoldas's method. The aim of the work was to achieve an effective control of the nanostructure properties and morphology of the final  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>. The influence of the reagent temperature during the hydrolysis was evaluated in case of water at 5 °C and 98 °C, and alkoxide at -18 °C and room temperature. Sol-gel transitions were performed at 120 °C and room temperature. All  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> samples were characterized by X-ray diffraction, nitrogen adsorption and thermal analysis. Our results showed that temperature of both water and alkoxide has not much influence on the nanostructure of the final  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>, thus giving a structure very similar to that of samples obtained by the reference method as long as the reaction temperature above 75 °C is reached soon enough. XRD characterization showed diffraction patterns corresponding to  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> for all samples. Also BET specific area values (253-280 m<sup>2</sup>/g) were similar to those obtained by Yoldas's original method. The temperature of the sol-gel transition does not affect the resulting sample structure, and crystalline boehmite particles were identified in all dried gels. We analyzed the reproducibility of the samples' structure by preparing different samples under identical conditions; we found that performing the sol-gel transition at 120 °C favors the production of more reproducible samples and also reduces significantly the time of the sol-gel reaction.

**Keywords :** nanostructure alumina, boehmite, sol-gel technique, N<sub>2</sub> adsorption/desorption isotherm, pore size distribution, BET area.

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