

Thermodynamics of Chlorination of Acid-Soluble Titanium Slag in Molten Salt for Preparation of TiCl_4

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Abstract : Chinese titanium iron ore reserves with high calcium and magnesium accounted for more than 90% of the total reserves, and acid-soluble titanium slag which is produced by titanium iron ore always used to produce titanium dioxide through sulphate process. To broad the application range of acid-soluble titanium slag, the feasibility and thermodynamics of chlorinated reaction for preparation TiCl_4 by titanium slag chlorination in molten salt were conducted in this paper. The analysis results show that TiCl_4 can be obtained by chlorinate the acid-dissolved titanium slag with carbon. Component's thermodynamics reaction trend is: $\text{CaO} > \text{MnO} > \text{FeO}(\text{FeCl}_2) > \text{MgO} > \text{V}_2\text{O}_5 > \text{Fe}_2\text{O}_3 > \text{FeO}(\text{FeCl}_3) > \text{TiO}_2 > \text{Al}_2\text{O}_3 > \text{SiO}_2$ in the standard state. Industrial experimental results are consistent with the thermodynamics analysis, the content of TiCl_4 is more than 98% in the production. Fe, Si, V, Al, and other impurity content can satisfy the requirements of production.

Keywords : thermodynamics, acid-soluble titanium slag, preparation of TiCl_4 , chlorination

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