

Mixed Micellization Study of Adiphenine Hydrochloride with 1-Decyl-3-Methylimidazolium Chloride

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Abstract : The mixed micellization of adiphenine hydrochloride (ADP) with 1-decyl-3-methylimidazolium chloride (C10mim.Cl), was investigated at different mole fractions and temperatures by surface tension measurements. The synergistic behavior (i.e., non-ideal behavior) for binary mixtures was explained by the deviation of critical micelle concentration (cmc) from ideal critical micelle concentration (cmc*), micellar mole fraction (X_{im}) from ideal micellar mole fraction (X_{iideal}), the values of interaction parameter (β) and activity coefficients (f_i) (for both mixed micelles and mixed monolayer). The excess free energy (ΔG_{ex}) for the ADP- C10mim.Cl binary mixtures explain the stability of mixed micelles in comparison to micelles of pure ADP and C10mim.Cl. Interfacial parameters, i.e., Gibbs surface excess (Γ_{max}), minimum head group area at air/ water interface (A_{min}), and free energy of micellization (ΔG_{0m}) were also evaluated for the systems.

Keywords : adiphenine hydrochloride, critical micelle concentration, interaction parameter, activity coefficient

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