Single Cu–N₄ Sites Enable Atomic Fe Clusters with High-Performance Oxygen Reduction Reaction

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Abstract : Atomically dispersed Fe–N₄ catalysts are proven as promising alternatives to commercial Pt/C for the oxygen reduction reaction. Most reported Fe–N₄ catalysts suffer from inferior O–O bond-breaking capability due to superoxo-like O₂ adsorption, though the isolated dual-atomic metal sites strategy is extensively adopted. Atomic Fe clusters hold greater promise for promoting O–O bond cleavage by forming peroxo-like O₂ adsorption. However, the excessively strong binding strength between Fe clusters and oxygenated intermediates sacrifices the activity. Here, we first report a Fex/Cu–N@CF catalyst with atomic Fe clusters functionalized by adjacent single Cu–N₄ sites and one a porous carbon nanofiber membrane. The theoretical calculation indicates that the single Cu–N₄ sites can modulate the electronic configuration of Fe clusters to reduce O₂* protonation reaction free energy, which ultimately enhances the electrocatalytic performance. Particularly, the Cu–N₄ sites can increase the overlaps between the d orbitals of Fe and p orbitals of O to accelerate O–O cleavage in OOH*. As a result, this unique atomic catalyst exhibits a half potential (E1/2) of 0.944 V in an alkaline medium exceeding that of commercial Pt/C, whereas acidic performance E1/2 = 0.815 V is comparable to Pt/C. This work shows the great potential of single atoms for improvements in atomic cluster catalysts.

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1

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