

## Optical and Luminescence Studies on Dy<sup>3+</sup> Singly Doped and Dy<sup>3+</sup>/Ce<sup>3+</sup> Co-doped Alumina Borosilicate Glasses for Photonics Device Application

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**Abstract :** We investigate the optical and photoluminescence properties from Dy<sup>3+</sup> singly doped and Dy<sup>3+</sup> co-doped with Ce<sup>3+</sup> alumino borosilicate glasses prepared using high temperature melt-quenching technique. The glass composition formula is 25SiO<sub>2</sub>-(40-x-y)B<sub>2</sub>O<sub>3</sub>-10Al<sub>2</sub>O<sub>3</sub>-15NaF-10ZnO-xDy<sub>2</sub>O<sub>3</sub> yCe<sub>2</sub>O<sub>3</sub> where, x = 0.5 mol% and y = 0, 0.1, and 0.5 mol%. The XRD study reveals the amorphous nature of both singly doped and co-doped glasses. Absorption study on Dy<sup>3+</sup> singly doped glass shows nearly twelve absorption peaks arising from the ground level of Dy<sup>3+</sup> ions (<sup>6</sup>H<sub>15/2</sub>) to various upper levels, and for Dy<sup>3+</sup>/Ce<sup>3+</sup> co-doped glasses, few of the transitions in the visible region are suppressed. The absorption band edge is shifted towards the higher wavelength region on increasing Ce<sup>3+</sup> concentration. The decrease in indirect energy bandgap and increase in Urbach energy of the prepared glasses is observed due to codoping with Ce<sup>3+</sup> ions. The photoluminescence studies on singly doped glass under 350 nm excitation showed three peaks at the blue (482 nm), yellow (575 nm), and red (663 nm) region. For codoped glasses, the emission peak at 403 nm is raised due to the 4d to 5f transition of Ce<sup>3+</sup> ions. Lifetime values (ms) of co-doped glass is found to be higher than singly doped glass. Under 350 nm excitation, CIE coordinates of the co-doped glasses moved towards the bright white light region. The correlated color temperature (CCT) values were obtained in the range 4500 - 4700 K. Thus, the prepared glasses can be used for photonics device applications.

**Keywords :** absorption spectra, borosilicate glasses, Ce<sup>3+</sup>, Dy<sup>3+</sup>, photoluminescence

**Conference Title :** ICGST 2022 : International Conference on Glass Science and Technology

**Conference Location :** New York, United States

**Conference Dates :** June 02-03, 2022