

## Adsorption of NO and NH<sub>3</sub> in MFI and H-ZSM5: Monte Carlo Simulation

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**Abstract :** Due to developing industries, the emission of pollutants such as NO<sub>x</sub>, SO<sub>x</sub>, and CO<sub>2</sub> are rapidly increased. Generally, NO<sub>x</sub> is attributed to the mono nitrogen oxides of NO and NO<sub>2</sub> that is one of the most important atmospheric contaminants. Hence, controlling the emission of nitrogen oxides is environmentally urgent. Selective catalytic reduction of NO<sub>x</sub> is one of the most common techniques for NO<sub>x</sub> removal in which zeolites have wide application due to their high performance. In zeolitic processes, the catalytic reaction occurs mostly in the pores. Therefore, investigation of the adsorption phenomena of the molecules in order to gain an insight and understand the catalytic cycle is of important. Hence, in current study, benefiting from molecular simulations, the adsorption phenomena in the nanocatalysts of SCR of NO<sub>x</sub> process was investigated in order to get a good insight of the catalysts' behavior. The effect of cation addition to the support in the catalysts' behavior through adsorption step was explored by Monte Carlo (MC) using Materials Studio Package. Simulation time of 1 ns accompanying 1 fs time step, COMPASS27 Force Field and the cut off radius of 12.5 Å was applied for performed runs. It was observed that the adsorption capacity increases in the presence of cations. The sorption isotherms demonstrated the behavior of type I isotherm categories and sorption capacity diminished with increase in temperature whereas an increase was observed at high pressures. Besides, NO sorption showed higher sorption capacity than NH<sub>3</sub> in H-ZSM5. In this respect, the energy distributions signified that the molecules could adsorb in just one sorption site at the catalyst and the sorption energy of NO was stronger than the NH<sub>3</sub> in H-ZSM5. Furthermore, the isosteric heat of sorption data showed nearly same values for the molecules; however, it indicated stronger interactions of NO molecules with H-ZSM5 zeolite compared to the isosteric heat of NH<sub>3</sub> which was low in value.

**Keywords :** Monte Carlo simulation, adsorption, NO<sub>x</sub>, ZSM5

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