

## Radical Degradation of Acetaminophen with Peroxymonosulfate-Based Oxidation Processes

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**Abstract :** Peroxymonosulfate (PMS)-based oxidation processes, as an alternative of hydrogen peroxide-based oxidation processes, are more and more popular because of reactive radical species ( $\text{SO}_4\cdot^-$ ,  $\text{OH}\cdot$ ) produced in systems. Magnetic nano-scaled particles  $\text{Fe}_3\text{O}_4$  and ferrous anion ( $\text{Fe}^{2+}$ ) were studied for the activation of PMS for degradation of acetaminophen (APAP) in water. The  $\text{Fe}_3\text{O}_4$  MNPs were found to effectively catalyze PMS for APAP and the reactions well followed a pseudo-first-order kinetics pattern ( $R^2 > 0.95$ ). While the degradation of APAP in PMS- $\text{Fe}^{2+}$  system proceeds through two stages: a fast stage and a much slower stage. Within 5 min, approximately 7% and 18% of 10 ppm APAP was accomplished by 0.2 mM PMS in  $\text{Fe}_3\text{O}_4$  (0.8g/L) and  $\text{Fe}^{2+}$  (0.1mM) activation process. However, as reaction proceed to 120 min, approximately 75% and 35% of APAP was removed in  $\text{Fe}_3\text{O}_4$  activation process and  $\text{Fe}^{2+}$  activation process, respectively. Within 120 min, the mineralization of APAP was about 7.5% and 5.0% (initial APAP of 10 ppm and  $[\text{PMS}]_0$  of 0.2 mM) in  $\text{Fe}_3\text{O}_4$ -PMS and  $\text{Fe}^{2+}$ -PMS system, while the mineralization could be greatly increased to about 31% and 40% as  $[\text{PMS}]_0$  increased to 2.0 mM in  $\text{Fe}_3\text{O}_4$ -PMS and  $\text{Fe}^{2+}$ -PMS system, respectively. At last, the production of reactive radical species were validated directly from Electron Paramagnetic Resonance (ESR) tests with 0.1 M 5,5-Dimethyl-1-pyrrolidine N-oxide (DMPO). Plausible mechanisms on the radical generation from  $\text{Fe}_3\text{O}_4$  and  $\text{Fe}^{2+}$  activation of PMS are proposed on the results of radical identification tests. The results demonstrated that  $\text{Fe}_3\text{O}_4$  MNPs activated PMS and  $\text{Fe}^{2+}$  anion activated PMS systems are promising technologies for water pollution caused by contaminants such as pharmaceutical.  $\text{Fe}_3\text{O}_4$ -PMS system is more suitable for slowly remediation, while  $\text{Fe}^{2+}$ -PMS system is more suitable for fast remediation.

**Keywords :** acetaminophen, peroxymonosulfate, radicals, Electron Paramagnetic Resonance (ESR)

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